Isolation and Characterization of an Eosinophilic GH 16 β-Agarase (AgaDL6) from an Agar-Degrading Marine Bacterium *Flammeovirga* sp. HQM9

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Introduction

Marine red macroalgae have received increasing attention as a sustainable resource for producing fuels and fine chemicals [1, 2]. Advantages of marine red macroalgae are fast growth, high yield, low lignin content or free of lignin and growth in the sea [3, 4]. The main carbohydrate of red macroalgae is agar which itself is composed of agarose and agarpectin. Agarose is a polysaccharide with alternating copolymers of 1-4-linked-3, 6-anhydro-α-L-galactose and 1-3-linked β-D-galactose [5]. Agarpectin has the same framework with some hydroxyl groups of 3,6-anhydro-α-L-galactose residues substituted by sulfate, methoxy, or pyruvate residues [6].

Agarose is the major component of red macroalgae. A suitable method of hydrolysis is necessary to obtain oligosaccharides from agarose. At present, oligosaccharides can be produced either by chemical hydrolysis or enzymatic hydrolysis. Enzymatic hydrolysis is the first choice because it has the advantages of mild reaction conditions, product control, environmental friendliness, and high efficiency compared with chemical hydrolysis processes [7]. Agarase, which degrades agarose to oligosaccharides or monosaccharides, is important in further fermentation and microbial utilization of this compound polysaccharide [8, 9]. Agarases can be classified into α-agarase (E.C. 3.2.1.158) and β-agarase (E.C. 3.2.1.81) according to their mode of action [2].

The α-agarases recognize and depolymerize the α-1,3 linkages to produce agaro-oligosaccharides with 3,6-anhydro-α-L-galactose at the reducing end. On the other hand, β-agarases recognize and cleave β-1,4 linkages to produce neoagarotetraose and neoagarohexaose as the final products. These characteristics make AgaDL6 a potentially valuable enzyme in the cosmetic, food, and pharmaceutical industries.

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produce neoagaro-oligosaccharides with D-galactose residues at the reducing ends [10, 11]. Compared with α-agarases, a large number of β-agarases have been found in many different genera, including Agarivorans [12], Cytophaga [13], Pseudomonas [14], Vibrio [15–17], Flammenceirgira [18], and Agarivorans [19, 20]. Based on amino acid sequence similarity, β-agarases have been classified into four distinct glycoside hydrolase (GH) families: GH16, GH50, GH86, and GH118 [21], which are recorded in the Carbohydrate-Active enZymes Database (http://www.cazy.org). β-agarases from the GH16, GH50, GH86 families contain two major conserved modules: one a catalytic glycoside hydrolase module which is responsible for hydrolyzing of glycoside linkage and the other a non-catalytic carbohydrate binding module (CBM), which helps enzymes bind substrates by forming a substrate-binding groove [22–24].

*Flammenceirgira* sp. HQM9, an agar-hydrolyzing marine bacterium, was isolated from the surfaces of red algae. We identified 34 complete coding sequences of agarases (all belonging to β-agarase) of *Flammenceirgira* sp. HQM9 genome by whole-genome sequencing [25]. Of these 34 agarases, 14 belonged to the GH-16 family, 6 belonged to GH-86, and only 2 belonged to GH-50. We predict that the remaining agarases belong to a new family.

Although a large number of agarases have been characterized in recent years, most are not resistant to acidic conditions and high temperature, which greatly limits their development and utilization. The β-agarase gene *AgaDL6*, being most acid- and thermostable, was cloned and characterized in this study as part of an effort to screen for more novel agarases with properties suitable for specific applications. In particular, agarase AgaDL6 had significant resistance to acidity.

### Materials and Methods

#### Bacterial Strains and Plasmids

*Flammenceirgira* sp. HQM9 was isolated from the surfaces of red algae and stored in our laboratory. It was cultivated using 2216E medium (2% NaCl, 0.3% MgCl₂·6H₂O, 0.6% MgSO₄·7H₂O, 0.1% (NH₄)₂SO₄, 0.02% NaHCO₃, 0.03% CaCl₂·2H₂O, 0.05% KCl, 0.042% KH₂PO₄, 0.005% NaBr, 0.002% SrCl₂·6H₂O, 0.1% yeast extract, 0.5% tryptone) supplemented with 1.5% agar. The pMAL-p5x vector was used for cloning and expression in *Escherichia coli* BL21 (DE3), which were routinely incubated at 37°C in Luria–Bertani (LB) broth. Ampicillin (100 μg/ml) was added when required.

#### Gene Cloning

Based on the results of genome analysis of *Flammenceirgira* sp. HQM9, the sequence of *AgaDL6* was obtained. After overnight culturing, the cell pellet of *Flammenceirgira* sp. HQM9 was harvested, and the genomic DNA was extracted. The *AgaDL6* gene was amplified by PCR using primers, forward (5-CATGCCATGGAT GAGATCATTTGTAAAGAAAAGTG-3, NcoI site is italicized) and reverse (5- ACGCGTCAGTTATTTATTATTTATTTAAGTGTTAAA TG-3, SalI site is italicized). The PCR product was digested with NcoI/SalI, and then transferred into the pMAL-p5x vector, which was also digested with NcoI/SalI.

#### Sequence Analysis and Classification of *AgaDL6*

We used the BLAST program (http://www.ncbi.nlm.nih.gov/BLAST) to search for sequence similarity and conserved domains. SignalP (version 4.1) (http://www.cbs.dtu.dk/services/SignalP/) was used for the prediction of the signal peptide sequence. Then, alignment of multiple sequences was performed through ClustalW (http://www.ch.embnet.org/software/ClustalW.html). The sequence analysis was conducted using DNAMAN (Version 6.0, Lynnon Biosoft Corporation, USA).

#### Expression and Purification of *AgaDL6*

*E. coli* BL21 harboring pMAL-p5x- *AgaDL6* was cultivated in an LB medium with ampicillin (100 μg/ml) at 37°C until a value of OD₆₀₀ of 0.6 was obtained. Subsequently, induction of the culture was performed using isopropyl β-D-1-thiogalactopyranoside (IPTG; 1 mM), followed by further culture of the transformant at 28°C for 10 h. Following centrifugation, the pellet of cells was collected and dissolved followed by sonication in 20 mM of PBS buffer. Further, the supernatant was subjected to purification with a PurKine MBP purification system (Abbkine, China). The fractions of the elution were collected and analyzed using SDS-PAGE.

#### Detection of Agarase Activity on Plate

To identify the plate-based activity assay of *AgaDL6*, 100 μl of purified *AgaDL6* was put in the holes of plates containing 2% agar, prepared in 20 mM Tris-HCl buffer (pH 7.0). The plates were incubated for 3 h at 40°C and then stained with Lugol’s iodine solution to check for reducing sugars. The clear zone around the hole was then visualized.

#### Assay of Enzymatic Activity of *AgaDL6*

The agarase activity was determined using the 3, 5-dinitrosalicylic acid (DNS) method [26]. Briefly, 30 μl of the enzymatic solution was admixed with 370 μl of 20 mM Tris-HCl buffer (pH 7.0) with the addition of agarose (1% w/v), followed by incubation for 30 min at 40°C. A volume of 400 μl of DNS reagent solution was added to the reaction mixture, which was next subjected to boiling for 10 min. After cooling, measurements of the absorbance were carried out at 540 nm. Calculations of the quantities of the reduced sugars were done using D-galactose as a standard. Finally, a unit of enzyme activity was designated as the amount of enzyme able to generate 1 μmol of D-galactose per minute through agarase hydrolysis.

Analysis of the Properties of Enzyme

The effect of temperature on AgaDL6 was assayed by incubating AgaDL6 with 1% (w/v) agarose at temperatures between 30°C and 70°C in 20 mM Tris-HCl (pH 7.0) for 30 min. The effect of pH on AgaDL6 was determined by incubating at 40°C for 30 min in buffers with different pHs: 20 mM sodium citrate buffer (pH 2–6), 20 mM MOPS buffer (pH 6–7), 20 mM Tris-HCl buffer (pH 7–9), and 20 mM borate buffer (pH 9–11). To measure the thermostability of AgaDL6, the enzyme was preincubated at various temperatures ranging from 30°C to 70°C for 2 h, and then the residual enzyme activity was measured. The pH stability of the AgaDL6 was evaluated by pre-incubating AgaDL6 at different pH values (2.0–10.0) at optimum temperatures for different times (from 12 to 48 h) and then measuring the residual enzyme activity. The relative activity was defined as a percentage with respect to the maximum agarase activity.

Effects of Various Metal Ions and Reagents on Agarase Activity

The effects of various metal ions, denaturants and chelators on AgaDL6 were investigated by measuring the enzyme activity supplemented with different concentrations (1 mM, 10 mM) of Na⁺, Mn²⁺, Mg²⁺, Zn²⁺, Ca²⁺, Ba²⁺, Cu²⁺, Co²⁺, Fe²⁺, EDTA, SDS and urea. All measurements were determined in triplicate. The relative activity was expressed as the percentage of activity determined with respect to the standard condition without metal ions, denaturants and chelators.

Identification of Hydrolysis Products

Thin-layer chromatography (TLC) was performed to identify hydrolysis products of AgaDL6 towards agar [25]. After incubating purified AgaDL6 with 1% agar at different times at optimum conditions, the hydrolysis reaction was stopped by heating in a boiling water bath for 10 min. The reaction mixture was spotted on silica gel 60 TLC plates (Merck, Germany). The plates were developed with n-butanol-acetic acid–water (2:1:1 by volume) solution. Spots were visualized by spraying with 20% H₂SO₄ in methanol and heating at 120°C for 2 min. Neoagarobiose (NA2), neoagarotetraose (NA4), neoagarohexaose (NA6), neoagaroctaose (NA8) and neoagaroctaose (NA10) were used as standards. The molecular mass of the enzymatic products was determined using a matrix-assisted laser desorption ionization time-of-flight (MALDI-TOF) mass spectrometer (AB Sciex, USA). The matrix for all samples was 2,5-dihydroxybenzoic acid (DHB) (10 mg/ml DHB in 30/70 acetonitrile/water with 0.1% TFA final concentration) and the ratio of the sample and matrix was 1:1. The measured data were analyzed and exported using Data Explorer Software 4.11 (AB Sciex).

Nucleotide Sequence Accession Number

The nucleotide sequence of AgaDL6 in strain Flammeovirga sp. HQM9 has been submitted to the GenBank database under accession number WP010523251. The draft genome sequence of HQM9 is available in GenBank under accession number AFBP00000000.

Results

Sequence Analysis

After PCR amplification using specific primers, the agarase gene was obtained and named AgaDL6. Sequence analysis showed that agarase gene AgaDL6 consists of 1,383 bp, encoding a protein of 461 amino acids with an estimated molecular mass of 52.8 kDa and a pI of 6.15. The search for conserved domains of AgaDL6 was performed according to the NCBI Conserved Domain Search. The result showed that AgaDL6 has two conserved domains: a glycoside hydrolase family 16 (GH16) catalytic module comprising 253 amino acids (Pro71–Arg323) in the N-terminal, and a carbohydrate-binding module 4_9 (CBM_4_9) comprising 72 amino acids (Gly333–Ala404) in the C-terminal (Fig. 1A).

Moreover, the amino acids of AgaDL6 showed high homology with β-agarases in the NCBI database: 86% to the β-agarase from Aquimarina agarilitytica (GenBank Accession No. WP_010177128), 61% to the β-agarase from Aquimarina agarilitytica (GenBank Accession No. WP_010177127), 55% to the β-agarase from Aquimarina agarizorans (GenBank Accession No. WP_010523252), and 55% to the β-agarase from Echinicola strongylcentrati (GenBank Accession No. WP_112786421). Multiple sequence alignments of AgaDL6 with known β-agarases were performed and the conserved catalytic residues were also analyzed (Fig. 1B). Based on the sequence homology between AgaDL6 and these members of GH16 family, we presumed that the amino acids of the active sites in AgaDL6 were Glu-182 and Glu-187 as the nucleophile and the acid/base, respectively [27]. The third catalytic site, the acidic amino acid residue Asp-185, may be related to maintenance of the charge environment during the interaction of the enzyme with the substrate [28].

Purification of AgaDL6 and Agarase Activity Assay

The β-agarase gene AgaDL6 was cloned into a pMal-p5x expression vector by fusion with a MBP at the N-terminal. The fusion protein pMal-AgaDL6 was purified to homogeneity by amylose resin under the induction of 1 mM IPTG. The purified AgaDL6 was observed as a single band on SDS-PAGE. Its molecular mass was estimated to be 95 kDa, matching the calculated size of AgaDL6 (52.8 kDa) with the MBP-tag (42.5 kDa) (Fig. 2A). In addition, a plate-based activity assay illustrated that purified AgaDL6 showed an obvious hydrolysis cycle on agar plates after a 3-h incubation (Fig. 2B), which initially demonstrated the agarase activity of AgaDL6.
Biochemical Analysis of Agarase

The effect of temperature on agarase activity of AgaDL6 was determined by incubating purified AgaDL6 at different temperatures (30°C to 70°C). The purified AgaDL6 exhibited maximum agarase activity at 50°C. AgaDL6 maintained more than 87% of its activity over a wide temperature range (40°C to 60°C). The enzymatic activity of AgaDL6 reduced sharply at 65°C, while 59% of its maximal activity was still maintained (Fig. 2D). Even after pre-incubation at 50°C or lower for 2 h, the agarase activity of AgaDL6 was maintained at almost 98% of its initial activity. The enzyme retained 84% of residual activities after incubation at 60°C for 2 h. When the pre-incubation temperature was raised to 65°C, the AgaDL6 activity was reduced to less than 25% of its initial activity.

In addition, as shown in Fig. 2C, the effect of pH on AgaDL6 activity was assayed by incubating AgaDL6 with agarose (1%, w/v) at different pH values (2 to 11). The results revealed that the optimal reaction pH of AgaDL6 was pH 3, and AgaDL6 exhibited strong stability across a wide range of pH (2–7). AgaDL6 maintained approximately 96% of its maximum activity across a range of pH (2 to 5) (Fig. 2C). Notably, AgaDL6 possesses excellent stability in the range of pH from 2.0 to 5.0 and retains 100% of activity after incubating for 24 h, saving approximately 90% activity even after 36 h incubation at the pH range mentioned above (Fig. 2E). Meanwhile, less than 40% of residual activity was retained after 24 h or longer incubation at pH values ranging from 7.0 to 10.0, proving that AgaDL6 is an acidic agarase.

Effects of Various Metal Ions and Reagents on the Activity of AgaDL6

The effects of different metal ions and reagents on AgaDL6 activity were investigated by using various metal ions and reagents at final concentrations of 1 or 10 mM. The AgaDL6 activities were stable in conditions with 10 mM macroelements contained in seawater, such as Na⁺, K⁺, and Mg²⁺ (Table 1). Ca²⁺ at 1 mM had no effect on AgaDL6 activity, but showed a slight positive effect at 10 mM. The
agarase AgaDL6 basically retained full activity after treatment with 1mM and 10 mM each of heavy metal ions such as Ba\(^{2+}\), Co\(^{2+}\), Fe\(^{3+}\), Zn\(^{2+}\), and Cu\(^{2+}\) ion. Meanwhile, the activity of AgaDL6 was slightly reduced by a high concentration of EDTA (10 mM), whereas a low concentration of EDTA, SDS and urea had a slight positive effect on the activity of AgaDL6.

**Oligosaccharide Degradation Patterns of AgaDL6**

An analysis of the hydrolysis pattern and products of agarase AgaDL6 was done to determine whether AgaDL6 was an endo- or exo-glycoside hydrolase, as well as to clarify its final hydrolysis products. A time course hydrolysis analysis was performed with the purified AgaDL6. The TLC analysis of the final products showed that AgaDL6 hydrolysed agarose into two predominate products, including neoagarotetraose (NA4), and neoagarohexaose (NA6) according to the standards (Fig. 3A). In the early stage of the reaction, agarose was rapidly degraded to NA4 and NA6. No neoagaro-oligosaccharides larger than NA6 were observed even after a long incubation of 24 h. Meanwhile, the results revealed that the amounts of NA4 and NA6 increased with the prolongation of the hydrolysis time, indicating that AgaDL6 has the characteristics of an exohydrolase.

We concluded that AgaDL6 is an exo-type β-agarase because of the NA4 and NA6 products. This speculation was further validated by TLC analysis using NA2, NA4, NA6, neoagaro-octaose (NA8) and neoagarodecaose (NA10) as a substrate (Fig. 3B). As shown in Fig. 3B, AgaDL6 could not hydrolyze NA2, NA4 and NA6. Whereas NA8 was completely hydrolyzed to NA4 and NA10 was digested to produce NA6 and NA4 by cleavage of the β-1,4 glycoside linkage. NA8 was the smallest oligosaccharide that AgaDL6 could hydrolyze and NA4 was the minimal oligosaccharide product.

To determine the exact masses of the hydrolysis products of AgaDL6, MALDI-TOF/TOF MS analysis was performed.
Table 1. Effects of various metal ions and chemical reagents on the activity of AgaDL6.

<table>
<thead>
<tr>
<th>Reagents</th>
<th>Relative activity (100%) of AgaDL6</th>
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</thead>
<tbody>
<tr>
<td></td>
<td>1 mM</td>
</tr>
<tr>
<td>Control*</td>
<td>100 ± 0.3</td>
</tr>
<tr>
<td>NaCl</td>
<td>103 ± 0.4</td>
</tr>
<tr>
<td>KCl</td>
<td>102 ± 0.3</td>
</tr>
<tr>
<td>CaCl$_2$</td>
<td>101 ± 0.4</td>
</tr>
<tr>
<td>BaCl$_2$</td>
<td>113 ± 0.3</td>
</tr>
<tr>
<td>MgCl$_2$</td>
<td>107 ± 0.5</td>
</tr>
<tr>
<td>CuCl$_2$</td>
<td>103 ± 0.1</td>
</tr>
<tr>
<td>ZnCl$_2$</td>
<td>102 ± 0.3</td>
</tr>
<tr>
<td>MnCl$_2$</td>
<td>91 ± 0.4</td>
</tr>
<tr>
<td>FeCl$_3$</td>
<td>99 ± 0.1</td>
</tr>
<tr>
<td>CoCl$_2$</td>
<td>97 ± 0.3</td>
</tr>
<tr>
<td>SDS</td>
<td>110 ± 0.5</td>
</tr>
<tr>
<td>EDTA</td>
<td>106 ± 0.4</td>
</tr>
<tr>
<td>Urea</td>
<td>108 ± 0.4</td>
</tr>
</tbody>
</table>

*The enzymatic activity with no metal ions, chelators or denaturants was set as 100%.

**Values represent the means ± standard deviation (SD) (n = 3).**

(Fig. 3C). MALDI-TOF mass spectra revealed that the major products have molecular ions at m/z of 653 (M+Na)$^+$, 669 (M+K)$^+$, 959 (M+Na)$^+$, and 975 (M + K)$^+$. These peaks were attributed to NA4 and NA6, respectively. Combining these results, the main end products of AgaDL6 were NA4 and NA6.

**Discussion**

Bacteria of the *Flammeovirga* genus have recently been identified from the surface of algae [18], deep-sea and coastal sediments [29], and the digestive tract of marine animals [30]. In our efforts to screen agar-hydrolyzing bacteria, a new agar-degrading bacterium, *Flammeovirga* sp. HQM9, was isolated from the surfaces of red algae. The genome of HQM9 was sequenced [25]. A new agarase gene AgaDL6 was screened and overexpressed in *E. coli*. The catalytic module of AgaDL6 contained three conserved characteristic motifs, DEFNY, EIDVLE and VDYVR at catalytically active sites and at calcium-binding regions [27], which were highly homologous to the conserved sequence of the GH16 catalytic motif [17].

The search for a conserved domain of AgaDL6 showed that the sequence in the N-terminal region was a beta-agarase domain belonging to the GH family 16; the protein did not contain the same C-terminal carbohydrate-binding module 6 (CBM6) module as previously reported for other GH16 members [31, 32]. All agarases reportedly contain a glycoside hydrolase domain, which is responsible for hydrolyzing the glycoside linkage and a carbohydrate-binding module (CBM), which helps enzymes bind substrates by forming a substrate-binding groove [23]. CBMs are generally classified into 71 families based on the similarity of amino acid sequences. At present, only CBM6 and CBM13 have been found in agarases [33]. Analysis using the CAZy and SMART databases suggested that AgaDL6 contained only one CBM_4_9 (Gly333 to Ala404) (Fig. 1A). This is significantly different from other GH16 members with CBM6 [33]. Although AgaDL6 belongs to a member of the GH16 family, our results show that it is different from the other members and thus a novel agarase belonging to the GH16 family obtained from bacteria of the genus *Flammeovirga*. The presence of CBM_4_9 in AgaDL6 is particularly interesting. The CBM likely functions as a dominant module that contributes to the special characteristics of AgaDL6.

Industrial processing requires an agarase with high stability and activity at temperatures above the gelling temperature of agar (about 40°C) [34]. Thermostability is important in the enzymatic conversion of agar or seaweed [28]. Since agarases come from the marine environment, a low temperature environment, most of them lose their activity at high temperatures, which greatly restricts the application of agarase [35, 36]. However, AgaDL6 exhibited thermostability during pre-incubation at 60°C for 2 h. These results indicate that agarase AgaDL6 has good thermostability. In this report, the stable and optimum temperatures of AgaDL6 are higher than the gelling temperature of agar.

AgaDL6 demonstrated maximum activity at pH 3.0. It is well known that natural seawater has a weak basic pH value so most agarases have been reported to exhibit optimum activity at a neutral [36, 37] or a weak basic pH [20, 38]. Rarely, agarase exhibits maximum activity under these conditions. Furthermore, AgaDL6 maintained approximately 95% of its maximum activity from pH 2 to 5. It is worth noting that AgaDL6 was extremely stable at these conditions, retaining 100% of activity even after 24 h of treatment at a pH range of pH 2.0–5.0, which reflects more acid resistance than other agarases.

The high degree of polymerization of agarose and the tendency to form a gel seriously hinder the hydrolysis of agarase [39]. Therefore, to efficiently obtain fermentable sugars from agarose, a chemical pretreatment such as acid
pretreatment can increase the enzymatic efficiency of agarose [40, 41]. Enzymatic hydrolysis combined with acid treatment can increase the yield of reducing sugar [42]. However, the high concentration of acid used in the acid pretreatment adversely affects the subsequent enzymatic reaction because known agarases are acid intolerant so an additional neutralization reaction is required. Similarly, the high salt concentration formed during the neutralization process also affects the subsequent enzymatic reaction [43]. These bottlenecks can be avoided by use of an acid-resistant acidic agarase, such as AgaDL6.

We also found that metals ions did not significantly activate or inhibit AgaDL6 activity. The activity of AgaDL6 was hardly influenced by 1 mM EDTA, and was only slightly inhibited by 10 mM EDTA, indicating that AgaDL6 is not a metal-ion dependent enzyme. Catalytic activity might not require the participation of divalent metal ions. This is a favorable property for industrial applications.

The hydrolysis pattern showed only two spots or peaks in both TLC and MALDI–TOF MS analysis: the amounts of two products increased with increasing reaction time, and no other oligosaccharides were observed during the enzymatic reaction. This hydrolysis pattern indicates there is no reaction intermediate such as NA8 or NA10 during the hydrolysis process, which is the classic hydrolysis mode of exo-agarases [20, 38]. In contrast, endo-agarase hydrolyzed agarose in a sort of random cleavage yielded oligosaccharides with different degrees of polymerization during the hydrolysis procedure [10]. As only two products were observed during enzymatic reaction and there was no change in the hydrolysis pattern, AgaDL6 was identified as an exo-type agarase.

AgaDL6 depolymerizes agarose into NA4 and NA6 as final products. This is distinct from most exo-agarases,
which produce only one type of oligosaccharide as the final product [44, 45]. The primary structure of the agarase determines its function and will ultimately affect its final products. Sequence analysis of AgaDL6 indicated that it has a special CBM_4_9 structure; therefore, the catalytic mechanism of AgaDL6 may be different from other agarases. Compared to an endolytic pattern, exo-type agarases reduce the complexity of oligosaccharide purification and the costs of product recovery by usually producing only two or fewer end product oligosaccharides. This is an obvious advantage in industrial applications.

In summary, AgaDL6 has these unique characteristics: outstanding acid resistance and stability, thermostability, high adaptability to high ion concentrations and simple hydrolysis products. Even after 24 h of treatment at a pH range of pH 2.0–5.0, AgaDL6 retained stability, unlike any other agarase. The high thermostability at temperatures higher than the gelatinization temperature has prospects for broad application in the enzymatic conversion of agar or seaweed. Catalytic activity might not require the participation of divalent metal ions. The reduced complexity of oligosaccharide purification reduces the costs of product recovery. These characteristics are important advantages for further industrial application compared with other agarases.

Acknowledgments

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Conflict of Interest

The authors have no financial conflicts of interest to declare.

References

An Eosinophilic GH16 β-Agarase, Carrying CBM_4_9


