Study of the Production of Alkaline Keratinases in Submerged Cultures as an Alternative for Solid Waste Treatment Generated in Leather Technology

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Introduction

Argentina has an economy strongly based on agriculture related industries. There are around 50 million livestock and more than 200 tanneries that process about 16 million of bovine hides per year. Although hair-saving unhairing processes reduce the organic load from beamhouse liquid effluent, a new solid residue called “hair waste” as sole sources of carbon and nitrogen. All the tested fungi grew on feather meal agar as well as in solid-state and submerged cultures, using a basal mineral medium and “hair waste” as sole sources of carbon and nitrogen. All the tested fungi grew on feather meal agar, but only three of them were capable of hydrolyzing keratin, producing clear zones. Among these strains, *P. lilacinum* produced the highest proteolytic and keratinolytic activities, both in solid-state and submerged fermentations. The medium composition and culture conditions for the keratinases production by *P. lilacinum* were optimized. Addition of glucose (5 g/l) and yeast extract (2.23 g/l) to the basal hair medium increased keratinases production. The optimum temperature and initial pH for the enzyme production were 28°C and 6.0, respectively. A beneficial effect was observed when the original concentration of four metal ions, present in the basal mineral medium, was reduced up to 1:10. The maximum yield of the enzyme was 15.96 U/ml in the optimal hair medium; this value was about 6.5-fold higher than the yield in the basal hair medium. These results suggest that keratinases from *P. lilacinum* can be useful for biotechnological purposes such as biodegradation (or bioconversion) of hair waste, leading to a reduction of the environmental pollution caused by leather technology with the concomitant production of proteolytic enzymes and protein hydrolyzates.

Keywords: Keratinases, hair waste, leather technology, *Purpureocillium lilacinum*

Six nonpathogenic fungal strains isolated from alkaline soils of Buenos Aires Province, Argentina (*Acremonium murorum*, *Aspergillus sidowii*, *Cladosporium cladosporoides*, *Neurospora tetrasperma*, *Purpureocillium lilacinum* (formerly *Paecilomyces lilacinus*), and *Westerdikella dispersa*) were tested for their ability to produce keratinolytic enzymes. Strains were grown on feather meal agar as well as in solid-state and submerged cultures, using a basal mineral medium and “hair waste” as sole sources of carbon and nitrogen. All the tested fungi grew on feather meal agar, but only three of them were capable of hydrolyzing keratin, producing clear zones. Among these strains, *P. lilacinum* produced the highest proteolytic and keratinolytic activities, both in solid-state and submerged fermentations. The medium composition and culture conditions for the keratinases production by *P. lilacinum* were optimized. Addition of glucose (5 g/l) and yeast extract (2.23 g/l) to the basal hair medium increased keratinases production. The optimum temperature and initial pH for the enzyme production were 28°C and 6.0, respectively. A beneficial effect was observed when the original concentration of four metal ions, present in the basal mineral medium, was reduced up to 1:10. The maximum yield of the enzyme was 15.96 U/ml in the optimal hair medium; this value was about 6.5-fold higher than the yield in the basal hair medium. These results suggest that keratinases from *P. lilacinum* can be useful for biotechnological purposes such as biodegradation (or bioconversion) of hair waste, leading to a reduction of the environmental pollution caused by leather technology with the concomitant production of proteolytic enzymes and protein hydrolyzates.

Keywords: Keratinases, hair waste, leather technology, *Purpureocillium lilacinum*
bonds. This characteristic provides to this protein mechanical stability and resistance to enzyme degradation by most common proteolytic enzymes, such as trypsin, pepsin, and papain [45].

Despite their high resistance, keratins are recycled in nature by keratinolytic microorganisms. Many fungi, especially those belonging to fungi imperfecti, have keratinolytic activity, including the following genera: Chrysosporium, Aspergillus, Alternaria, Cladosporium, Fusarium, Geomyces, Paecilomyces, Scopulariopsis, Penicillium, and Doratomyces [16, 22, 34, 47]. The enzymes involved in keratin degradation are called keratinases; they act as proteases but differ from the non-keratinolytic ones for their ability to degrade insoluble keratin [14].

Understandably, hair waste locks up a great deal of potentially useful proteinaceous material that could be beneficially harnessed for different purposes. Biotransformation of this waste implies considering it as raw material instead of the present idea of disposability. Thus, the hair would be the substrate onto which a microorganism (in our case, a fungus) would act (and grow), giving rise to a partially degraded organic material with high nitrogen content. This new biomaterial could show different potential uses, such as a highly digestible protein source for animal feeding, a raw material in the fertilizing or chemical industry, or a new type of hair peptone for microbial culture medium preparation.

The aim of this work was to select the most suitable microorganism with keratinolytic activity between 6 locally isolated fungal strains. The ability to grow and metabolize hair waste, in both solid-state and submerged culture conditions, was evaluated. In addition, the present work describes the effect of temperature, initial pH, glucose concentration, salts concentrations, and different nitrogen sources in order to enhance keratinases production by the fungus selected. In order to identify the type of proteases produced, some biochemical properties of the extracellular P. lilacinum keratinases were determined.

Materials and Methods

Microorganisms

Six nonpathogenic fungal strains were used in this work. They were locally isolated from alkaline forest soils and belong to the Spegazzini Institute fungal culture collection (La Plata National University, Argentina): Acremonium murorum (Corda) Gams var. murorum LPS # 57, Aspergillus sidoides LPS # 931, Cladosporium cladosporoides LPS # 953, Neurospora tetrasperma LPS # 837, Purpureocillium lilacinum (formerly Paecilomyces lilacinus) (Thom) Samson LPS # 876, and Westerdiikella dispersa LPS # 834. Stock cultures of all strains were kept on potato dextrose agar slants under mineral oil at 4°C and subcultured periodically.

Qualitative Test of Keratinolytic Activity on Feather Meal Agar Plates

The ability to degrade keratin was tested on feather meal agar (FMA). This medium was composed of (g/l) feather meal, 15; NaCl, 0.5; K2HPO4, 0.3; KH2PO4, 0.4, and agar, 15 (pH 7.2) [45]. The strains were punctually inoculated and incubated at 28°C for 15 days. The capability to degrade keratin was determined according to the presence or absence of a hydrolysis halo. Those strains presenting a hydrolysis halo were preselected for solid-state and submerged fermentations [45].

Inoculum Preparation for Solid-State and Submerged Fermentations

The strains were streaked on potato dextrose agar dishes and incubated at 28°C for 10 days. After this period of time, spores were harvested by flooding the agar dish with 10 ml of 0.05% Tween 80 sterile solution and by gently stirring the surface with a sterile magnetic bar. The concentration of spore suspension was determined in a Neubauer chamber.

Culture Medium

A basal mineral medium described by Galarza et al. [19] was used for both solid-state and submerged fermentations. The composition of the mineral medium was (g/l) NaH2PO4, 0.496; K2HPO4, 2.486; FeCl3·6H2O, 0.016; ZnCl2, 0.013; MgCl2·6H2O, 0.010, and CaCl2·0.0011 (pH 7.0). Hair waste, obtained from a local tannery, was washed extensively with tap water and dried at 60°C for 2 days, and kept at room temperature until used. In all cases, it was used as the sole carbon and nitrogen sources.

Culture Conditions

Solid-state fermentation (SSF) was carried out in autoclaved Petri dishes containing 3 g of hair waste impregnated with 10 ml of basal mineral medium already inoculated with a spore concentration of 2 × 103 per ml, and incubated at 28°C in a humid chamber for 28 days. Two dishes were withdrawn every 2 days for enzyme extraction. The extraction was carried out by adding 30 ml of 0.5 M NaCl solution to each plate under stirring for 15 min. The crude extract obtained was vacuum filtered through a 0.45 μm cellulose membrane in an ice-water bath and kept at −20°C until used for protein and activity determination [20].

Submerged fermentations (SF) were carried out in 500 ml Erlenmeyer flasks containing 200 ml of the basal mineral medium and 2 g of hair waste. The medium containing basal mineral medium plus hair waste will be called basal hair medium. Flasks were autoclaved at 121°C for 15 min and then inoculated with 2 × 104 conidia per ml. Cultures were performed in an orbital shaker at 200 rpm and 28°C for 17 days, withdrawing 5 ml of the culture
Samples were centrifuged at 1,500 × g for 15 min, at 4°C. The supernatant was used for pH determination and then kept at −20°C until used for quantification of protein content and enzyme activities.

Scanning Electron Microscopy (SEM)

To characterize the degradation of hair waste by *P. lilacinum*, digested and undigested hair waste samples were freeze-dried and then coated with gold palladium. SEM was accomplished using a Jeol JSM-840 microscope at an accelerating voltage of 25 kV.

Protein Determination

Protein concentration was determined by Bradford’s method [7] using bovine albumin fraction V (Sigma) as a standard.

Determination of Enzyme Activities

Proteolytic activity. Protease activity was measured as described by Liggieri et al. [32] using azocasein as the substrate, but with some modifications. A 0.1 ml aliquot of the enzyme preparation, suitably diluted, was mixed with 0.250 ml of 0.1 M Tris-HCl buffer (pH 9.0) containing 1% (w/v) of azocasein, and incubated for 30 min at 37°C. The reaction was stopped by the addition of 1.0 ml of trichloroacetic acid (10% w/v). The mixture was allowed to stand at room temperature for 15 min and then centrifuged at 5,000 × g for 10 min. A blank was performed with 0.1 ml of heat-inactivated enzyme. The 1 ml of 1 M NaOH was added to 1 ml of the supernatant and the absorbance was measured at 440 nm. One unit of protease activity (U) was defined as the amount of enzyme that, under the experimental conditions, causes an increase of 0.1 units in the absorbance at 440 nm per minute.

Keratinolytic activity. Keratinase activity was determined as described by Joshi et al. [27] using azokeratin as substrate. Substrate solution containing 30 mg of azokeratin and 0.5 ml of 0.1 M Tris-HCl buffer (pH 9.0) was stirred for 15 min until the azokeratin was completely suspended. An appropriate dilution of azokeratin was completely suspended. An appropriate dilution of 0.1M Tris-HCl buffer (pH 9.0) was stirred for 15min until the substrate solution containing 30mg of azokeratin and 0.8ml of 30min at 37°C. The reaction was stopped by the addition of 1.0 ml of 1M NaOH was added to 1ml of the supernatant and the absorbance was measured at 440 nm. One unit of keratinolytic activity (U) was defined as the amount of enzyme that, under the above-mentioned reaction conditions, causes an increase of 0.01 units in the absorbance at 440 nm per minute.

Optimization of the Medium Components for Keratinase Production

Effect of metal ions present in the basal mineral medium. Several experiments were done in order to study the effect of concentrations of four metal ions (Ca²⁺, Fe³⁺, Zn²⁺, and Mg²⁺) present in the basal mineral medium on keratinase production. Diluted solutions equivalent to 1:2, 1:4, 1:10, 1:25, and 1:50 of the original solution were tested. The initial pH of the medium was set at 7.0.

Effect of glucose. In order to study the effect of glucose on the production of keratinolytic enzymes, two concentrations of this sugar were tested: 5 and 10 g/l. The initial pH of the medium was set at 7.0, and the four metal ions present in the basal mineral medium were diluted up to 1:10.

Effect of nitrogen sources. The influence of various nitrogen sources on the keratinase production was studied. Yeast extract (2.23 g/l), urea (0.40 g/l), (NH₄)₂SO₄ (1.10 g/l), and KNO₃ (1.69 g/l) were added to the basal hair medium. The initial pH of the medium was set at 7.0 and the four metal ions present in the basal mineral medium were diluted up to 1:10.

Effect of environmental conditions on production of keratinases.

Temperature (28°C and 37°C) and initial pH (6.0, 7.0, and 8.0) were evaluated using the basal hair medium.

Biochemical Characterization of the Extracellular Keratinases

All experiments referring to partial biochemical characterization of the extracellular keratinases were performed using an enzyme preparation obtained from the day of maximum enzyme production in SF.

Detection of keratinolytic activity on polyacrylamide gels (casein and keratin zymograms). Casein and keratin zymographies were performed to test the protease and keratinase activity, respectively. Zymographies were performed in conjunction with SDS-PAGE according to the method of García-Carreño et al. [21] with slight modification. SDS-PAGE was performed as described by Laemmli [30], using 5% (w/v) stacking gel and 12% (w/v) separating gel. In the case of SDS-PAGE, samples heated and not heated were used, and for zymography the sample was not heated prior to electrophoresis. For zymography, after electrophoresis, the gel was submerged in 100 mM Tris-HCl buffer (pH 9.0) (buffer A) containing 2.5% Triton X-100 for 60 min, with constant agitation in order to remove the SDS. Triton X-100 was then removed by washing the gel three times with buffer A. A portion of the gel was incubated with 1% (w/v) casein or soluble keratin in buffer A at 28°C for 30 and 90 min, respectively. Finally, gels were stained with Coomassie Brilliant Blue R-250 for zymography analysis. The development of clear zones on the blue background of the gels indicated the presence of protease or keratinase activity. The molecular mass markers used were phosphorylase b (97 kDa); albumin (66 kDa); ovalbumin (45 kDa); carbonic anhydrase (30 kDa); trypsin inhibitor (20.1 kDa); and α-lactalbumin (14.4 kDa).

Effects of protease inhibitors and metal ions. The effects of the following inhibitors of protease activity were investigated:
phenylmethylsulfonyl fluoride (PMSF, 1 mM), iodoacetate (10 mM), ethylenediaminetetraacetate (5 mM), 1,10-phenanthroline (1 mM), and Pepstatin A (chlorambucil, 100 µg/ml). The enzyme preparation was preincubated in the presence of each inhibitor for 1 h at room temperature (20°C), and the protease activity remaining during subsequent assay was expressed as a percent of the control value with enzyme not exposed to inhibitor.

The effects of different metal ions (at a concentration of 1 mM) on protease activity were studied by addition of the cations Ca²⁺, Mg²⁺, Zn²⁺, and Hg²⁺ to the enzyme solution, followed by incubation for 1 h at room temperature. The protease activity remaining upon subsequent assay was expressed as a percent of the control value with enzyme not exposed to cations.

Statistical Analysis

All analyses were performed at least in triplicate. The data were expressed as means ± standard deviations. Univariate analysis of variance (ANOVA) was employed on the data for protease activity in the study production at different pH, temperatures, complex and inorganic nitrogen sources, and carbon supplement as well as in the biochemical characterization, and tested for their significance.

Results and Discussion

Qualitative Test in Feather Meal Agar (FMA)

In a recent screening made by the Spegazzini Institute on 69 fungal strains isolated from local alkaline soils, 32 proved to produce alkaline keratinase activity [15]. From these, 6 nonpathogenic strains were chosen because of their comparatively high keratinolytic activity (A. murorum, A. sidowii, C. cladosporoides, N. tetrasperma, P. lilacinum, and W. dispersa) and were tested for growth and feather meal degradation on FMA. All the strains tested were able to grow, but only 3 of them (A. murorum, C. cladosporoides, and P. lilacinum) produced a hydrolysis halo around the colony after 15 days of culture, a fact ascribed to the production of extracellular keratinases (Fig. 1). A similar behavior has been reported in a screening of dermatophytes, where among the 16 fungal strains tested, only Trichophyton verrucosum showed tiny fungal colonies surrounded by a wide clear zone of solubilized keratin. Wawrzkiewicz et al. [51] attributed this observation to an active secretion of keratin-decomposing enzymes to the medium. Sangali and Brandelli [46] also utilized this technique in order to screen several microorganisms for their ability to hydrolyze keratin. According to the results presented in this qualitative test, A. murorum, C. cladosporoides, and P. lilacinum were selected for further experiments.

Solid State Fermentations

Solid-state fermentations have been exploited for the production of a number of fungal enzymes owing to a number of economic advantages over conventional submerged cultures. Therefore, and according to the results of plate screening using FMA, A. murorum, C. cladosporoides, and P. lilacinum were tested for growth and enzyme production in SSF using hair waste as substrate. Among these strains, the highest extracellular proteases production, measured by the azocasein assay, was obtained with P. lilacinum (Fig. 2A). The fungus produced a large amount of extracellular proteases, detected after 3 days, and reached the maximum level after 10 days. The proteolytic activity remained almost constant for around 8 days, showing a constant enzyme production during that period of time. Many strains of fungi and bacteria species producing protease activity (keratinase activity) under solid-state fermentation have been described [12, 29, 40], but just one of them used hair waste as substrate; in that case, this substrate was just used to make a screening for potential keratinolytic fungi, but neither production of the proteases nor their characterization was done [15].

Submerged Fermentation

Although a number of keratin wastes have been used for the production of keratinases, there are only a few reports of the use of bovine hair waste for this purpose using submerged fermentation [33, 38, 43, 48, 52] Therefore, it was decided to test the three fungal strains (A. murorum, C. cladosporoides, and P. lilacinum) for enzyme production in SF using hair waste as substrate in order to compare the fungal behavior and the enzyme production between these
Although the strains were able to grow, *P. lilacinum* reached the maximum production of enzyme activity after 5 days (Fig. 2B). During the whole fermentation, the pH of the medium increased because of the ammonia produced by the deamination of peptides and aminoacids derived from keratin solubilization [41].

On the other hand, soluble protein concentration showed a similar behavior to that of the proteolytic activity, except for the fact that the maximum peak was reached a few days later (Fig. 3). Since keratin is used as the medium component, keratinase production is accompanied by a subsequent degradation of keratin substrate, leading to an increase in soluble protein concentration. The increment of soluble protein has been reported as a measure for keratin degradation [5]. The decrease of soluble protein may be due to an increment of the proteolysis degree, thus releasing peptides that were not detected by Bradford’s reagent.

Comparison of agar plate screening (FMA) with SSF and SF showed that the results of enzyme activity obtained were not parallel. This might be attributed to the use of different keratin sources with singular structures. Feather meal is mainly composed of a highly denatured hard keratin (β-keratin) and could be hydrolyzed more easily than hair waste (α-keratin) [8]. Therefore, it can be proposed that the semi-quantitative test in FMA gives a preliminary idea of the keratinolytic capacity of a microorganism, and this property must or should be confirmed by quantitative assessment of the enzymatic production from SSF or SF using hair waste.

Accordingly to the results presented up to this point, *P. lilacinum* was selected for further studies of the optimization of the extracellular protease production. The extent of degradation of hair waste by *P. lilacinum* was confirmed by SEM. Figs. 4A and 4B show SEM images of uninoculated hair fibers after a hair-saving unhairing process using sodium sulfite/lime as unhairing agent, and hairs fibers after 5 days of SF, respectively. A considerable degradation and disorganization of the fiber due to the fungal attack during SF was observed and also fungus...
aggregates with an extracellular matrix adhered to the degraded surfaces (Fig. 4C).

Culture Conditions for Keratinolytic Enzyme Production

The determination of optimal growth conditions yielding the highest keratinases production was achieved by analyzing the influence of several factors. Fig. 5A shows the time courses of keratinolytic, proteolytic, and soluble protein production of *P. lilacinum* in the basal hair medium. Maximum protease and keratinase enzyme production of 2.46 U/ml and 25 U/ml, respectively, was achieved at around 111–117 h of incubation. It can be seen in Fig. 5A that both activities increased together, keeping a constant keratinolytic:caseinolytic (K:C) activity ratio of 11.32 ± 1.06. This fact confirms the previous assumption concerning the utilization of the proteolytic activity as a direct estimation of keratinolytic activity. As was mentioned before, it was observed that the production of soluble protein had the same tendency as the production of keratinolytic enzyme. The extracellular pH of the medium increased constantly during the cultivation.

Temperature is an important environmental factor affecting the growth and production of metabolites by microorganisms [6]. The optimum temperature for the production of keratinolytic protease by *P. lilacinum* was 28°C. At 37°C, both the growth and protease activity of *P. lilacinum* were very poor (Table 1). This temperature optima for maximum enzyme production is almost similar to those reported for *Vibrio* sp. kr2 [45], *Lysobacter* sp. [2], *Streptomyces* sp. [12], and *Penicillium* sp. [13], which showed an optimum temperature for keratinases production ranging from 20°C to 30°C, whereas *Trichophyton* sp. and *Aspergillus nidulans* have optimum temperatures near 40°C [3, 44].

In order to study the effect of reducing the concentration of metal ions present in the basal mineral medium, several submerged cultures were carried out using diluted concentrations...
of four metal ions present in the basal mineral medium (Ca\(^{2+}\), Fe\(^{3+}\), Zn\(^{2+}\), and Mg\(^{2+}\)). It is clear from the results presented in Table 2 that the enzyme production was considerably enhanced when the concentration of these four elements was reduced to 1:10. Jeong et al. [26] studied the influence of different inorganic salts on keratinases production, demonstrating that maximal production was found in a medium containing metal ions like Ca\(^{2+}\) and Mg\(^{2+}\). Studies concerning proteopectinase production by Geotrichum klebahnii [17] showed that a reduction of the combination of yeast extract and glucose supplemented to 5g/l of glucose. Table 3 shows that the addition of yeast extract and (NH\(_4\))\(_2\)SO\(_4\) increased keratinase production 1.56-fold and 1.13-fold, respectively. However, the addition of other co-nitrogen sources, such as urea or KNO\(_3\), repressed keratinase production. It is worth to mention that the combination of yeast extract and glucose supplemented to the basal hair waste medium showed a positive cumulative influence (U\(_c\)/ml ~ 16) on enzyme production by this organism. Similar results were reported by Hossain et al. [24], where a combination of NH\(_4\)Cl (0.1% (w/v)) and molasses (1% (w/v)) showed a positive influence on keratinase production by B. licheniformis MZK-3.

Finally, the effects of the medium’s initial pH on keratinase production were investigated in a pH range of 6.0–8.0. P. lilacinum was able to grow and produce proteases in the range studied, and the optimum pH for the production was 6.0, where the enzyme production was increased 2-fold relative to the basal mineral medium (4.23 U\(_c\)/ml) (Table 1). This result was similar to Pseudomonas sp. KP-364 (pH 6.5) and Ferrovibacterium penneri (pH 6.3) [9, 18], but not with those reported for the production of keratinase from Paracoccus sp. WJ-98 (pH 7.5 [31]) or for the production

### Table 1. Effects of temperature (°C) and initial pH of the medium on keratinases production by P. lilacinum after 115 h of incubation.

<table>
<thead>
<tr>
<th>T (°C)</th>
<th>Protease activity (U(_c)/ml)</th>
</tr>
</thead>
<tbody>
<tr>
<td>28</td>
<td>2.46 ± 0.04</td>
</tr>
<tr>
<td>37</td>
<td>0.31 ± 0.05</td>
</tr>
</tbody>
</table>

<table>
<thead>
<tr>
<th>Initial pH</th>
<th>Protease activity (U(_c)/ml)</th>
</tr>
</thead>
<tbody>
<tr>
<td>6</td>
<td>4.23 ± 0.30</td>
</tr>
<tr>
<td>7</td>
<td>2.46 ± 0.04</td>
</tr>
<tr>
<td>8</td>
<td>2.16 ± 0.04</td>
</tr>
</tbody>
</table>

Values are the average of three independent experiments ± standard deviations.

### Table 2. Influence of various dilutions of the metal ions present in the mineral medium on enzyme production.

<table>
<thead>
<tr>
<th>Dilution of metal ions</th>
<th>Protease activity (U(_c)/ml)</th>
</tr>
</thead>
<tbody>
<tr>
<td>1:1</td>
<td>2.46 ± 0.04</td>
</tr>
<tr>
<td>1:2</td>
<td>2.56 ± 0.10</td>
</tr>
<tr>
<td>1:4</td>
<td>2.71 ± 0.17</td>
</tr>
<tr>
<td>1:10</td>
<td>3.10 ± 0.07</td>
</tr>
<tr>
<td>1:25</td>
<td>2.10 ± 0.12</td>
</tr>
<tr>
<td>1:50</td>
<td>2.02 ± 0.13</td>
</tr>
</tbody>
</table>

Values are the average of three independent experiments ± standard deviations.

### Table 3. Influence of glucose concentration and of co-nitrogen sources on keratinolytic enzyme production by P. lilacinum in basal hair medium.

<table>
<thead>
<tr>
<th>Co-nitrogen sources</th>
<th>Protease activity (U(_c)/ml)</th>
</tr>
</thead>
<tbody>
<tr>
<td>None</td>
<td>3.10 ± 0.07</td>
</tr>
<tr>
<td>5 g/l</td>
<td>10.46 ± 0.51</td>
</tr>
<tr>
<td>10 g/l</td>
<td>10.68 ± 0.61</td>
</tr>
<tr>
<td>Yeast extract</td>
<td>15.96 ± 0.56</td>
</tr>
<tr>
<td>Urea</td>
<td>2.09 ± 0.39</td>
</tr>
<tr>
<td>(NH(_4))(_2)SO(_4)</td>
<td>11.58 ± 0.22</td>
</tr>
<tr>
<td>KNO(_3)</td>
<td>3.57 ± 0.60</td>
</tr>
</tbody>
</table>

Values are the average of three independent experiments ± standard deviations.
of a serine protease from B. licheniformis MZK-3 (pH 8.0 [24]).

As result of this study, an optimal condition for the enzyme production was established. The optimal hair medium contained 0.5% (w/v) glucose, 0.22% (w/v) yeast extract, a dilution of 1:10 of the original solution of metal ions, and 1% (w/v) of hair waste (pH 6.0 and 28°C), respectively.

Fig. 5B shows the time course of keratinolytic enzyme production of P. lilacinum in the optimal hair medium. The level of keratinolytic enzyme produced by P. lilacinum increased considerably (6.5-fold) and reached a maximum (15.96 U/ml) at 86 h, and then decreased slightly.

Biochemical Characterization of the Extracellular Keratinases

Detection of keratinolytic activity on polyacrylamide gels (casein and keratin zymograms). In this study, P. lilacinum LPS #876 was found to produce at least two proteases, but just one of them has keratinolytic activity as revealed by the casein and keratin zymograms (Fig. 6). These results are in line with those reported by Huang et al. [25], where just one protease was produced by B. pumilis with dehairing capabilities. Nevertheless, several works reported the production of more than one extracellular protease. Agrebi et al. [1] reported the production of at least seven proteases by B. pumilis A 26; Xie et al. [52] reported the presence of five proteases in the culture supernatant of Streptomyces sp. strain 16 growing on human foot skin medium; and Mazotto et al. [35] reported the production of seven proteases in the culture supernatant of human hair medium supplemented with yeast extract by B. subtilis AMR, from which three resulted to be keratinases accordingly to keratin zymograms. Interestingly, when the supernatant was submitted to SDS-PAGE and to the corresponding zymography, it could be seen that enzyme activities were not affected by the presence of β-mercaptoethanol or by SDS (present in SDS-PAGE solutions), and the apparent molecular mass seems to be high, but when the enzyme was heated, its apparent molecular mass resulted to be near 37 kDa. Moreover, when the supernatant was subjected to a Superdex 75 chromatography, the active fraction collected was one that corresponds to a molecular mass near to 35 kDa (data not shown), suggesting that the presence of SDS could induce conformational changes that do not affect enzyme activity. At high temperatures, denaturation “opens” the polypeptide chain, rearranging the pattern of intra and intermolecular interactions within the protein and with the solvent [4]. Similar behavior was reported by Muga et al. [37] in relation to the interaction of SDS with β-Galactosidase.

Influence of inhibitor on enzyme activity. Proteases can be classified by their inhibition profile. The effects of several of these inhibitors, such as PMSF, EDTA, pepstatin A, etc., were investigated (Table 4). The keratinases secreted by P. lilacinum were slightly inhibited by EDTA, pepstatin A, iodoacetate, or 1,10-phenantroline. PMSF caused strong inhibition of the enzyme activity (93%) suggesting that the enzymes secreted belong to the class of serine proteases. This type of inhibition was reported for proteases from

![Fig. 6. SDS-PAGE and zymography analysis of the crude extract.](image)

Lane 1: molecular marker. Lane 2: heated crude extract supernatant. Lane 3: non-heated crude extract supernatant. Lane 4: casein zymogram, and Lane 5: keratin zymogram.

Table 4. Effects of protease inhibitors and metal ions on the enzyme activity of keratinase from P. lilacinum.

<table>
<thead>
<tr>
<th>Chemical</th>
<th>Concentration</th>
<th>Residual activity (%)</th>
</tr>
</thead>
<tbody>
<tr>
<td>None</td>
<td>100</td>
<td></td>
</tr>
<tr>
<td>Inhibitor</td>
<td></td>
<td></td>
</tr>
<tr>
<td>PMSF</td>
<td>2 mM</td>
<td>7.0 ± 0.0</td>
</tr>
<tr>
<td>Iodoacetate</td>
<td>10 mM</td>
<td>95.1 ± 4.7</td>
</tr>
<tr>
<td>EDTA</td>
<td>5 mM</td>
<td>99.6 ± 6.3</td>
</tr>
<tr>
<td>1,10-Phenantroline</td>
<td>1 mM</td>
<td>100 ± 0.2</td>
</tr>
<tr>
<td>Pepstatin A</td>
<td>100 µg/ml</td>
<td>87.5 ± 5.5</td>
</tr>
<tr>
<td>Metal ion</td>
<td></td>
<td></td>
</tr>
<tr>
<td>Mg²⁺</td>
<td>1 mM</td>
<td>105.0 ± 1.2</td>
</tr>
<tr>
<td>Zn²⁺</td>
<td>1 mM</td>
<td>92.8 ± 1.4</td>
</tr>
<tr>
<td>Ca²⁺</td>
<td>1 mM</td>
<td>102.9 ± 0.5</td>
</tr>
<tr>
<td>Hg²⁺</td>
<td>1 mM</td>
<td>6.0 ± 0.6</td>
</tr>
</tbody>
</table>

All the chemicals tested were preincubated with the enzyme at room temperature (25°C), for 1 h, and then the residual enzyme activity was measured. Values are the average of three independent experiments ± standard deviations.
**Vibrio fluvialis** [49], *Trichophyton vanbreuseghemii* [36], and *Bacillus* sp. JB9 [28].

**Influence of metal ions on enzyme activity.** As can be seen in Table 4, in the presence of Ca$^{2+}$, a moderate increase of enzyme activity was observed, but in the presence of Zn$^{2+}$, a slight inhibition of the activity was observed. It has been reported that this metal ion has an inhibitory effect on alkaline proteases [23]. Hg$^{2+}$ strongly inhibited the proteolytic activity (94% of inhibition); this metal ion is recognized as an oxidant agent of thiol groups, and the enzyme inhibition by this ion suggests the presence of an important SH group (such as free cysteine) at or near the active site [11].

Through the results obtained in this work, it was possible to determine that *P. lilacinum* strain, isolated from alkaline soils from Buenos Aires Province, produced extracellular keratinolytic activity when cultivated with hair waste as a sole sources of carbon and nitrogen both in SSF and SF. The strain presents potential uses in biotechnological processes involved in the degradation of keratinous wastes from several industries, such as that of leather, contributing to the reduction in disposal problems. At the same time, it produces compounds resulting from keratin hydrolysis that can be used for the production of biodegradable films, glue and protein hydrolysates, balanced food, and fertilizers, etc. In addition, the keratinolytic enzymes involved in this biotransformation could be considered as a not less important product. After recovery, and even as crude extracts, they could find application in the food and leather industries as well as in the manufacture of textiles, cosmetics, etc. On the other hand, the optimal medium composition improved the keratinases production by about 6.5-fold. The conditions could be applied for a large-scale fermentation system.

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**References**


